

**PVCDROM**

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Instructions

1. Introduction
2. Properties of Sunlight
3. PN Junction
4. Solar Cell Operation
5. Design of Silicon Cells
6. Manufacturing Si Cells
7. Modules and Arrays
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**9. Material Properties**

 General Properties of Silicon  
Optical Properties of Silicon

Bi2S3

CdS

CdSe

CuO

FeS2

Mg2Si

SnS

**ZnSe**

CuInSe2

CuS

MnS

TiS2

**11. Appendices**

Korean Version

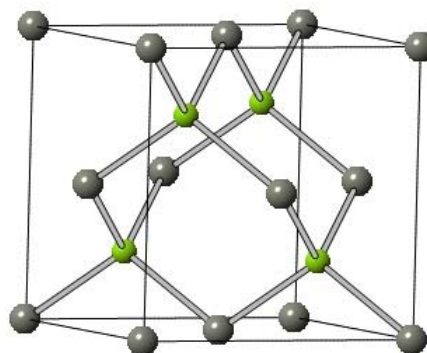
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[< SnS](#)
**ZnSe**
[CuInSe2 >](#)
**Material: Zinc Selenide (ZnSe)**
**Basic Info**

ZnSe is a light yellow binary solid compound that rarely occurs in nature. It can be made in hexagonal and cubic crystal structures. Zinc selenide is used as a blue light source in light-emitting diodes and diode lasers. It is also used as infrared laser gain medium and as an infrared optical material. ZnSe activated with tellurium is used in x-ray and gamma ray detectors as a scintillator. [9]

**Crystal Structure**


	Label	Elmt	Fractional Coordinates			Orthogonal Coordinates		
			x	y	z	xor[Å]	yor[Å]	zor[Å]
1.	T2	Se	0.2500	0.2500	0.2500	-0.863	-1.331	-1.873
2.	T2	Se	0.2500	0.7500	0.7500	0.100	-3.980	-4.722
3.	T2	Se	0.7500	0.7500	0.2500	-2.657	-1.164	-5.453
4.	T2	Se	0.7500	0.2500	0.7500	-3.482	-4.171	-2.935
5.	T1	Zn	0.0000	0.0000	0.0000	0.000	0.000	-0.000
6.	T1	Zn	0.0000	0.5000	0.5000	0.963	-2.649	-2.849
7.	T1	Zn	0.5000	0.5000	0.0000	-1.794	0.167	-3.580
8.	T1	Zn	0.5000	0.0000	0.5000	-2.620	-2.841	-1.063
9.	T1	Zn	1.0000	0.0000	0.0000	-5.376	-0.025	-1.793
10.	T1	Zn	1.0000	0.5000	0.5000	-4.414	-2.674	-4.642
11.	T1	Zn	0.0000	1.0000	0.0000	1.788	0.358	-5.366
12.	T1	Zn	0.5000	1.0000	0.5000	-0.831	-2.482	-6.429
13.	T1	Zn	1.0000	1.0000	0.0000	-3.588	0.333	-7.159
14.	T1	Zn	0.0000	0.0000	1.0000	0.137	-5.656	-0.332
15.	T1	Zn	0.5000	0.5000	1.0000	-1.657	-5.490	-3.912
16.	T1	Zn	1.0000	0.0000	1.0000	-5.239	-5.681	-2.125
17.	T1	Zn	0.0000	1.0000	1.0000	1.925	-5.298	-5.698
18.	T1	Zn	1.0000	1.0000	1.0000	-3.451	-5.323	-7.491

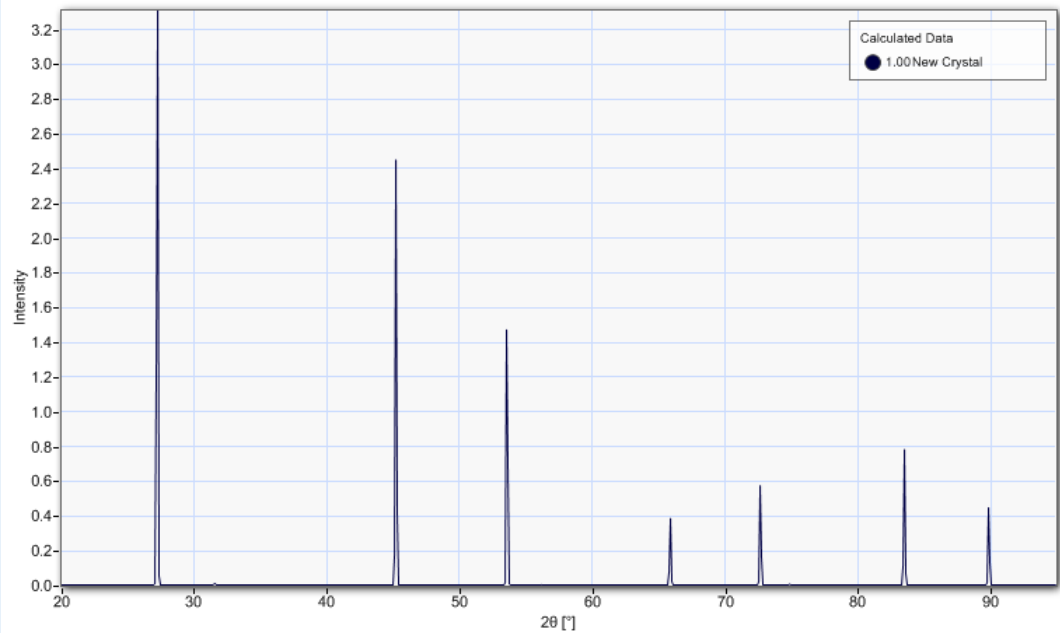
 Theoretical diffraction data using a Cu K $\alpha$  monochromatic source.

(m is the multiplicity and N is the maximum number of flexions)

ref no.	h	k	l	d(hkl)	2-Theta	Intensity	I/Imax	m	N
[ 1 ]	1	1	1	3.27219	27.2296	3.56799e-001	100.0	8	3
[ 2 ]	0	0	2	2.83380	31.5438	1.03393e-003	0.3	6	4
[ 3 ]	0	2	2	2.00380	45.2123	2.62259e-001	73.5	12	8
[ 4 ]	1	1	3	1.70885	53.5827	1.57080e-001	44.0	24	11

[\[ + \] Feedback](#)

[ 5 ]	2	2	2	1.63610	56.1706	3.14514e-004	0.1	8	12
[ 6 ]	0	0	4	1.41690	65.8603	4.10933e-002	11.5	6	16
[ 7 ]	1	3	3	1.30024	72.6539	6.12994e-002	17.2	24	19
[ 8 ]	0	2	4	1.26731	74.8585	6.43795e-004	0.2	24	20
[ 9 ]	2	2	4	1.15689	83.4861	8.34838e-002	23.4	24	24
[ 10 ]	1	1	5	1.09073	89.8496	4.77530e-002	13.4	32	27
[ 11 ]	0	4	4	1.00190	100.4906	3.01446e-002	8.4	12	32



## PV Applications

ZnSe thin films are prepared primarily by molecular beam epitaxy, chemical vapor deposition, chemical vapor deposition, and vacuum evaporation. Electrodeposition is a simple, low cost and a viable method for producing good quality films.[2]

ZnSe films prepared by chemical vapor deposition have a buffer layer that has reached total area efficiencies of up to 9.6% (under AM 1.5 illumination), an open circuit voltage of 482 mV, a short circuit current of 31.0 mA/cm<sup>2</sup> and a fill factor reaching 64%. [8]

## Basic Parameters at 300 K

Crystal structure:	Sphalerite	[1]
Group of symmetry:	F-43m	[1]
Number of atoms in 1 cm <sup>3</sup> :	4.39*10 <sup>26</sup>	[1]
Unit cell volume:	182.05 Å <sup>3</sup>	[1]
Atoms per unit cell:	8	[1]
Debye temperature:	339(2) K	[7]
Density:	5.266 g/cm <sup>3</sup>	[1]
Dielectric constant (static):	8.6	[10]
Dielectric constant (high frequency):	5.7	[10]
Effective electron masses:	(0.16 ± 0.01)m <sub>e</sub>	[3]
Effective hole masses:	0.75 m <sub>o</sub>	[7]
Lattice constant:	a = 5.667 Å	[1]
Optical phonon energy (longitudinal):	0.0314 eV	[5]
Conductivity:	n-type	[3]

## Temperature Dependences

Graph of electron concentration vs temperature may be found in M. Aven, High Electron Mobility in Zinc Selenide Through Low Temperature Annealing. J. Appl. Phys. 42, 1204 (1971); doi: 10.1063/1.1660167 [5]

## Donors and Acceptors

Donors:	Al, Cl, Ga, In, F, Br	[3],[4]
Acceptors:	Cu, Ag, Sb	[4]

Ionization energy of shallow donors [7]

E<sub>d</sub>(Li<sub>I</sub>) 15(1) meV T= 4.2 K

E<sub>d</sub>(Na<sub>I</sub>) 16(1) meV

$E_d(\text{Al})$	26.3 meV
$E_d(\text{Ga})$	27 meV
$E_d(\text{In})$	28.1 meV
$E_d(\text{F})$	29.3 meV
$E_d(\text{Cl})$	26.1 meV
$E_d(\text{I})$	30.4 meV

Ionization energy of shallow acceptors

$E_a(\text{Li})$	118(2) meV	$T = 4.2 \text{ K}$
$E_a(\text{Na})$	98(2) meV	
$E_a(\text{K})$	94(2) meV	
$E_a(\text{N})$	112 meV	$T = 4.2 \text{ K}$
$E_a(\text{P})$	80...92 meV	$T = 4.2 \text{ K}$
$E_a(\text{As})$	125 meV	$T = 77 \text{ K}$
$E_a(\text{Sb})$	69 meV	$T = 30 \text{ K}$
$E_a(\text{Rb})$	89(2) meV	$T = 4.2 \text{ K}$
$E_a(\text{Cs})$	74(2) meV	
$E_a(\text{O})$	80 meV	$T = 4 \text{ K}$
$E_a(\text{V}_{\text{Zn}})$	218 meV	$T = 4 \text{ K}$

## Electrical Properties

### Basic Parameters of Electrical Properties

[2]	Energy gap:		2.81 eV	
	Energy spin-orbital splitting:	$\Delta_0 (\Gamma_{8v} - \Gamma_{7v})$	0.42 eV	$T=295 \text{ K}$ [7]
		$\Delta_1 (\Gamma_{4,5v} - \Gamma_{6v})$	0.20 eV	$T=300 \text{ K}$ [7]
[5]	Donor concentration:		$10^{16} \text{ cm}^{-3}$	
	Carrier mobility:		$\mu_n = \text{up to } 400 \text{ cm}^2/\text{Vs}$	$T=300\text{K}$ [7]
			$\mu_p = 110 \text{ cm}^2/\text{Vs}$	$T=300\text{K}$ [7]
	Intrinsic resistivity:		$\sim 10^{12} \Omega \text{ cm}$	[6]

### Mobility and Hall Effect

Hall mobility:	$530 \text{ cm}^2 / \text{V*s}$ ( $T=300 \text{ K}$ )	[6]
	$12,000 \text{ cm}^2 / \text{V*s}$ ( $T=60 \text{ K}$ )	[6]
Absorption coefficient:	$10^4 \text{ cm}^{-1}$	[2]

Mobilities and mobility ratios as well as a graph of electron hall mobility vs temperature may be found in:

M. Aven, High Electron Mobility in Zinc Selenide Through Low Temperature Annealing. J. Appl. Phys. 42, 1204 (1971); doi: 10.1063/1.1660167 [5]

## Optical properties

Refractive index: Graph of refractive index and absorption index vs. photon energy may be found in Madelung, O. (2004). *Semiconductors: Data handbook*. (3rd ed., pp. 736-757). Springer. [7]

## Thermal properties

Coefficient of linear thermal expansion:	$\alpha = 7.4 \cdot 10^{-6} \text{ K}^{-1}$	[7]
Heat capacity:	$C_p = 51.88 \text{ J/Mol} \cdot \text{K}$	[7]
Thermal conductivity	$\kappa = 0.19 \text{ W K}^{-1} \text{ cm}^{-1} \text{ T}=300\text{K}$	[7]

Graphs of ZnSe's thermal properties may be found in Madelung, O. (2004). *Semiconductors: Data handbook*. (3rd ed., pp. 736-757). Springer. [7]

## Mechanical properties, elastic constants, lattice vibrations

### Basic Parameters

Bulk modulus:	62.4(7) GPa	[7]
Density:	5.266 g/cm <sup>3</sup>	[7]

### Elastic Constants

Elastic Constants:	$C_{11} = 90.3(19) \text{ GPa}$	[7]
	$C_{12} = 53.6(23) \text{ GPa}$	[7]
	$C_{44} = 39.4(12) \text{ GPa}$	[7]

### Phonon Frequencies

$\nu_{LO}(\Gamma)$	7.59 THz	T=300 K	[7]
$\nu_{TO}(\Gamma)$	6.39 THz	T=300 K	

### Phonon Energies

$H\nu_{LO}(\Gamma_1)$	30.99 meV	[7]
$H\nu_{TO}(\Gamma_{15})$	25.17 meV	
$H\nu_{LA}(\Gamma)$	19.8 meV	
$H\nu_{TA}(\Gamma)$	8.0 meV	
$H\nu_{LO}(X)$	27.64 meV	
$H\nu_{TO}$	25.54 meV	
$H\nu_{LA}$	23.55 meV	
$H\nu_{LO}(L)$	27.77 meV	
$H\nu_{TO}$	25.54 meV	
$H\nu(W_3)$	24.9 meV	
$H\nu(W_1)$	18.59 meV	
$H\nu(W'_2)$	11.53 meV	
$H\nu(W''_2)$	26.53 meV	
$H\nu(W'_4)$	14.26 meV	
$H\nu(W''_4)$	24.61 meV	

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The development of these pages on photovoltaic materials' properties was carried out at the University of Utah primarily by undergraduate students Jeff Provost and Carina Hahn working with Prof. Mike Scarpulla. Caitlin Arndt, Christian Robert, Katie Furse, Jash Sayani, and Liz Lund also contributed. The work was fully supported by the US National Science Foundation under the Materials World Network program award 1008302. These pages are a work in progress and we solicit input from knowledgeable parties around the world for more accurate or additional information. Contact [earthabundantpv@eng.utah.edu](mailto:earthabundantpv@eng.utah.edu) with such suggestions. Neither the University of Utah nor the NSF guarantee the accuracy of these values.



« 8n8

CuIn8a2 »

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